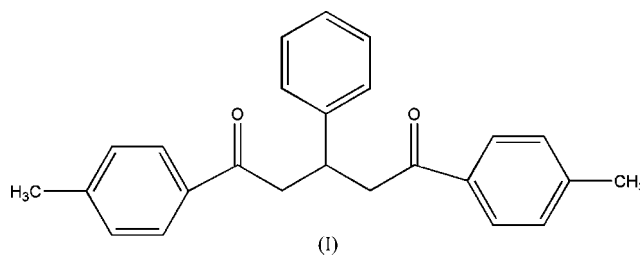
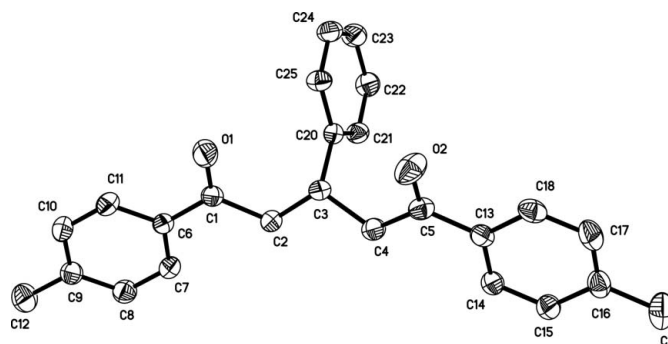


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Key indicatorsSingle-crystal X-ray study
 $T = 298$ K
Mean $\sigma(\text{C}-\text{C}) = 0.006$ Å
 R factor = 0.042
 wR factor = 0.126
Data-to-parameter ratio = 8.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**1,5-Bis(4-methylphenyl)-3-phenylpentane-
1,5-dione**The title compound, $\text{C}_{25}\text{H}_{24}\text{O}_2$, has been synthesized by the
reaction of benzaldehyde with 4-methylacetophenone and
 NaOH . The bond lengths and angles show normal values. The
crystal packing exhibits no significantly short intermolecular
contacts.Received 28 April 2006
Accepted 16 May 2006**Comment**Recently, solvent-free organic reactions (Toda, 1995; Loupy,
2000; Cave *et al.*, 2001) have attracted great attention due to
the increasing concern for the protection of the environment
(DeSimone, 2002; Tanaka, 2003; Tanaka & Toda, 2000;
Metzger, 1998). 1,5-Diketones are important synthetic inter-
mediates and starting materials in the synthesis of many
heterocyclic compounds (Hirsch & Bailey, 1978; Krohnke,
1976). In a continuation of previous work on the synthesis of
1,5-diketones (Constable *et al.*, 1998; Fuchigami *et al.*, 1986),
we present a new compound, 1,5-bis(4-methylphenyl)-3-
phenylpentane-1,5-dione, (I), synthesized under solvent-free
conditions.In the molecule (Fig. 1), the bond lengths and angles are
normal and correspond to those observed in 1,3,5-triphenyl-
pentane-1,5-dione (Das *et al.*, 1994). The crystal packing
demonstrates no significantly short intermolecular contacts.**Figure 1**
View of (I) with the atomic numbering and displacement ellipsoids drawn
at the 30% probability level. H atoms have been omitted for clarity.

Experimental

4-Methylacetophenone (0.75 g, 6.25 mmol), freshly distilled benzaldehyde (0.33 g, 3.125 mmol) and NaOH (0.25 g, 6.25 mmol) were aggregated with a glass paddle in an open flask. The resulting mixture was washed with water several times to remove NaOH and was recrystallized from ethanol, affording the title compound as a crystalline solid.

Crystal data

$C_{25}H_{24}O_2$	$Z = 4$
$M_r = 356.44$	$D_x = 1.222 \text{ Mg m}^{-3}$
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 5.715 (3) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$b = 15.848 (8) \text{ \AA}$	$T = 298 (2) \text{ K}$
$c = 21.386 (11) \text{ \AA}$	Block, colourless
$V = 1936.8 (17) \text{ \AA}^3$	$0.23 \times 0.18 \times 0.15 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	10270 measured reflections
φ and ω scans	1998 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1166 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.983, T_{\max} = 0.989$	$R_{\text{int}} = 0.076$
	$\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.056P)^2 + 0.0975P]$
$R[F^2 > 2\sigma(F^2)] = 0.042$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.126$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.01$	$\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$
1998 reflections	$\Delta\rho_{\text{min}} = -0.15 \text{ e \AA}^{-3}$
244 parameters	
H-atom parameters constrained	

All H atoms were positioned geometrically, with C–H = 0.93–0.98 Å, and refined as riding, with $U_{\text{iso}}(\text{H}) = U_{\text{eq}}(\text{C})$. Due to the

absence of any significant anomalous scatterers in the molecule, the 1409 Friedel pairs were merged before the final refinement.

Data collection: SMART (Siemens, 1996); cell refinement: SMART; data reduction: SAINT (Siemens, 1996); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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References

Cave, G. W. V., Raston, C. L. & Scott, J. L. (2001). *Chem. Commun.* pp. 2159–2169.

Constable, E. C., Neuburger, M., Smith, D. R. & Zehnder, M. (1998). *Inorg. Chim. Acta*, **275–276**, 359–365.

Das, G. C., Hursthouse, M. B., Malik, K. M. A., Rahman, M. M., Rahman, M. T. & Olsson, T. (1994). *J. Chem. Crystallogr.* **24**, 511–515.

DeSimone, J. M. (2002). *Science*, **297**, 799–803.

Fuchigami, T., Awata, T., Nonaka, T. & Baizer, M. M. (1986). *Bull. Chem. Soc. Jpn*, **59**, 2873–2879.

Hirsch, S. S. & Bailey, W. J. (1978). *J. Org. Chem.* **43**, 4090–4094.

Krohnke, F. (1976). *Synthesis*, pp. 1–24.

Loupy, P. (2000). *Top. Curr. Chem.* **206**, 153–207.

Metzger, J. O. (1998). *Angew. Chem. Int. Ed. Engl.* **37**, 2975–2978.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.

Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Tanaka, K. (2003). Solvent-free Organic Synthesis. Weinheim: Wiley-VCH.

Tanaka, K. & Toda, F. (2000). *Chem. Rev.* **100**, 1025–1074.

Toda, F. (1995). *Acc. Chem. Res.* **28**, 480–486.